UNIVERSITY OF MUMBAI



Syllabus for the F.Y.B.Sc. Program: B.Sc. Course: Biochemistry

(Credit Based Semester and Grading System with effect from the academic year 2017 – 2018)

Preamble

Biochemistry is central to all areas of the "biological" and "life" science. It aims to provide an understanding of every aspect of the structure and function of living things at cellular level. Being an interdisciplinary subject it is spanning a wide range of areas from microbiology to plant and animal sciences to pathology of diseases and nutrition.

The impact of studies in biochemistry on modern life is enormous. Therefore, the syllabus is structured to touch upon broad base at the beginning. Unique physical and chemical characteristics of water enable it to function in ways essential to human and other life processes due to its structure and composition. Life on Earth began more than 3 billion years ago, evolving from the most basic of microbes into a dazzling array of complexity over time, which makes it necessary to study the origin of life and evolution of a modern species over span of years. After an in-depth understanding of how the first cells originated, students are introduced to detailed structural organization of basic unit of a living system "The Cell". Biomolecules are the basic and important constituents of a living system. Hence, it is mandatory to study structure, occurrence and functions of large biomolecules like carbohydrates, lipids and proteins along with nucleic acids. In order to prepare the students for detailed course in Applied Nutrition in the higher education, the syllabus is made to understand human nutrition and its significance. In order to understand the biological processes occurring in the living body, processes as digestion, absorption, respiration and excretion are necessary to be studied. As stated earlier, life evolved from a small microbe, it is our aim to study living microscopic size organisms which includebacteria, fungi, protozoa and special type of microorganisms called extremophiles

Objectives of the first year of the course

- Develop an adequate background to enable the first year students to study more advanced biochemistry topics.
- Acquaint the learners with the unique properties of the universal solvent water, essential for life processes.
- Understand the life constituting bio molecules: proteins, carbohydrates, lipids, nucleic acids.
- Familiarize the learners about the origin of life and take them through the process of evolution.
- Focus on Cell as the basic unit of life which is the center for all biochemical processes.
- Familiarize the learners to the world of microorganisms which exist as independent cellular units.
- Develop an interest in the learner in nutrition for sustaining life, and physiology and functioning of life systems.
- Appreciate the importance of the broad spectrum of biochemistry.
- Provide familiarity with basic biochemistry laboratory techniques.
- Develop the practical skills of students to enhance their observational skills and to use these skills for problem solving.

F. Y. B. Sc. Biochemistry Syllabus

Credit Based Semester and Grading System

To be implemented from the academic year 2017 - 2018

Semester I

Course Code	Unit	Topics	Credits	Lectures	
	Biomolecules and Nutrition				
USBCH101	Ι	Water	2	15	
	II	Amino acids and proteins		15	
		Carbohydrates		15	
Introduction cell biology, Physiology and Microbiology					
USBCH102	I	Origin of life and formation of cells	2	15	
	II	The cell wall, cell membrane, cell organelles and cell division		15	
		Microbiology I		15	
USBCHP01		Practicals based on both courses in theory - USBCH101 and USBCH102	2	2	

Semester II

Course Code	Unit	Topics	Credits	Lectures
		Biomolecules and Nutrition		
USBCH201	I	Lipids	2	15
	II	Nucleic acids		15
		Nutrition		15
Introduction cell biology, Physiology and Microbiology				
USBCH202	I	Physiology of digestion and absorption	2	15
	II	Physiology of respiration and excretion		15
		Microbiology II		15
USBCHP02		Practicals based on both courses in theory - USBCH201 and USBCH202	2	2

F.Y.B.Sc. Biochemistry Syllabus Restructured for Credit Based and Grading System to be implemented from the Academic year 2017 - 2018 Semester I

USBCH101 - Bio molecules and Nutrition

Course Code	Title	Credits
USBCH101	Biomolecules and Nutrition	2 Credits (45 lectures)
Unit I: Water	15 Lectures	
 1.1 Water: Its (surface te properties) 1.1.1 Entropy ar 1.1.2 Effect of ne 1.1.3 Weak inter 		
1.2Solutions1.2.1Concepts	of mole, molar, molar equivalent and normal, Dalton	
1.3Ionization1.3.1pH: pH sc1.3.2Weak acid1.3.3Buffers - c(No derival)	n of water, weak acids and weak bases ale, H ⁺ and OH ⁻ concentrations Is and bases and their dissociation constants Ka & Kb lefinition, action, physiological buffers - phosphate and carbonate itions. Only simple problems on solutions)	
Unit II: Amino a	cids and proteins	15 Lectures
2.1 Amino aci 2.1.1 Amino aci 2.1.2 Detailed c 2.1.3 Physical p melting po Chemical Edman's	ids d structure - D & L forms of all 20 amino acids lassification based on polarity, essential and non essential amino acid properties: zwitter ions, pl of amino acids amino acids as ampholytes, bint, optical rotation, UV absorption and chemical properties: reactions of amino acids with Ninhydrin, Sanger's reagent, reagent and Dansyl chloride	
2.2 Peptides 2.2.1. ASBC - A 2.2.2 Primary st 2.2.3 Secondar 2.2.4 Tertiary(r 2.2.5 Protein de	and Proteins PS classification on the basis of shape and function ructure - Formation and characterization of the peptide bond y structure - Alpha helix and beta sheet hyoglobin) and Quaternary(hemoglobin)structures - an introduction maturation	
Unit III: Carbohy	vdrates.	15 Lectures
3.1 Definition, (mono, oli	Classification, and functions of carbohydrates to polysaccharides)	
 3.2 Monosace 3.2.1 Classificat 3.2.2 Occurrenc and ribose 3.2.3 Properties a) Physica b) Chemica i) oxida (with ii) redu (with iii) Osa iv) Orci 	charides on in terms of aldoses and ketoses e,structures and significance of glucose, fructose, galactose, mannose, - isomerism D & L, optical; epimers : anomers al reactions - ation to produce aldonic. aldaric and uronic acids respect to glucose); cing action in boiling alkali, enediol formation respect to glucose and fructose) zone formation (with respect to glucose and fructose). nol (with respect to ribose)	

 3.3 <i>Disaccharides</i> 3.3.1 Occurrence and structure of maltose , lactose and sucrose 3.3.2 Formation of glycosidic bonds 	
 3.4 Polysaccharides 3.4.1 Classification based on function. storage and structure a) Composition: homo & hetero. with examples b) Storage : starch and glycogen - action of amylase on starch c) Structural: cellulose, chitin 	

USBCH102 - Introduction to Cell biology, Physiology and Microbiology

Course Code	Title	Credits
USBCH102 Introduction to Cell biology, Physiology and Microbiology		2 Credits (45 lectures)
Unit I: Origin of	Life & Formation of cells	15 Lectures
1.1 Big bang the hypothesis, endosymbio	eory, Theories on the origin of life: Abiogenesis, Heterotroph RNA world, protein world, Miller's experiment, Formation of the first cell, nt theory	
1.2 Evolution - E Gene mutati evidences: F geographic generations Hardy-Wein	Darwinian theory, Modern synthetic theory of evolution and its factors: ons(recombination), heredity, natural selection and isolation Biological Fossil record,chemical and anatomicalsimilarities of related life forms, distribution of related species, genetic changes in living organisms over and Mechanism of evolution, Gene flow and genetic drift, berg principle	
Unit II: The cell-	15 Lectures	
 2.1 Structural 2.1.1 Prokaryotic 2.2 Cell wall str Cytoskelete 	organization of cells c, Eukaryotic (plant & animal) and yeast cells - a comparative overview ucture (plant), cell membrane (fluid mosaic model) on: microtubules & microfilaments	
 2.3 Cell organe 2.3.1 Mitochondi 2.3.2 Chloroplas other plast 2.3.3 Ribosome: 2.3.4 Peroxisom Lysosome 2.3.5 Nucleus: S nuclearmat 2.4 Mitosis an 	elles: ion: Organization & function of the mitochondria, mitochondrial genome t: Structure and function of the chloroplast. the chloroplast genome, ids <i>ER</i> : <i>Golgi</i> Structure & Function of Ribosome, ER, Golgi apparatus e & Lysosome: Peroxisome function & assembly (in brief) and structure and function tructure & function of the nucleus, nuclear envelope, nuclear pores, rix and Nucleolus d Meiosis	

Uni	15 Lectures	
3.1	Historical background (contributions of Leeuwenhoek. Pasteur ,etc) and general characteristics (size, shape and structure) of Bacteria	
3.2	<i>Microbial Taxonomy</i> : Microbial species and strains. classification of bacteria based on morphology (shape and flagella), staining reaction. nutrition and extreme environment	
3.3	Bacterial cell wall: Structure and function, components of peptidoglycan frame work (structures of NAG and NAMA not necessary)	
3.4	An introduction to extremophiles: thermophiles, psychrophiles, halophiles, magnetotactic, radiation resistant - examples with their application	
3.5	Staining methods (principles of staining & types or stains) and microscopic identification of bacteria	

SEMESTER I - USBCHP01

PRACTICAL – I

- Preparation & Standardisation of laboratory reagents Primary standards - 0.1N oxalic acid Secondary standards - 0.1N NaOH, 0.1N HCI
- 2. Preparation of buffers acetate and phosphate
- 3. Determination of pKa of acetic acid
- 4. Qualitative tests for Carbohydrates -
 - Monosaccharides (glucose and fructose),
 - Disaccharides (lactose ,maltose and sucrose)
 - Polysaccharides (starch and dextrin)
 - unknown
- 5. Qualitative test for amino acids
- 6. Effect of heat, organic solvents and ammonium sulphate on proteins

DEMONSTRATION EXPERIMENT

pH meter - working of a pH meter

PRACTICAL – II

- 1. Effect of isotonic, hypertonic and hypotonic solutions on cells onion peel
- 2. Staining of bacterial yeast cells (negative staining)
- 3. Staining techniques-
 - gram staining,
 - endospore,
 - capsule and
 - lipids
- 4. Permanent slides/ diagrams or electron micrograph of organelles-nucleus, mitochondria and chloroplast
- 5. Study of stages of mitosis using onion root tips
- 6. Permanent slides of mitosis and meiosis

DEMONSTRATION EXPERIMENT

1. Microscopy – study of a compound microscope

45 Lectures hours

45 Lectures hours

Semester II USBCH201- Bio molecules and Nutrition

Course Code	Title	Credits
USBCH201	Biomolecules and Nutrition	2 Credits (45 lectures)
linit I · Linide		15 Lectures
	descritication (Discrict) and functions of Linida	
1.1 Definition, (classification (Bloor S) and functions of Lipids	
1.2 Fatty' acids 1.2.1 Classificat	s and Triacylglvcerol on &Chemistry,	
Saturated IUPAC nar Omega - 3	fatty acids - classification of C2 to C20: even carbon: Common and nes. Unsaturated fatty acids MUFA, PUFA (2.3.4 double bonds) .6.9 fatty acids. Triacyl glycerol - simple and mixed - names and structure	
Saponifica Action of h Rancidity Iodine nur	ation, Iodination, Ozonolysis, Auto-oxidation, neat on glycerol and choline, Definition & significance - Acid number, Saponification number, nber, Reichert - Meissel number	
1.3 Compound Functions Phosphosp (gluco & ga	l Lipids of glycerophospholipids (PE.PC.PL) phingolipids (ceramide, sphingomyelin), Glycolipids /Cerebrosides alactocerebrosides)	
1.4 Steroids Ch	nolesterol structure and biochemical significance.	
Unit II : Nucleic	Acids	15 Lectures
2.1 Structure - nucleotides with its sho	Purine & Pyrimidine bases, ribose, deoxyribose, nucleosides and (ATP, CTP, GTP, TTP, UTP) Formation of polynucleotide strand rthand Representation	
2.2 RNAs (vari account, t	ous types in prokaryotes and eukaryotes) mRNA & rRNA - general RNA - clover leaf model, Ribozymes	
 2.3 DNA 2.3.1 Physical e (chemical e 2.3.2 Physical p (Viscosity, denaturati 2.3.3 Reactions 	vidence of DNA helical structure. Chargaff's rules evidence), Watson-Crick model of DNA & its features roperties of DNA - Effect of heat on physical properties of DNA buoyant density, UV absorption), Hypochromism, hyperchromism, on of DNA. of nucleic acids (with DPA and Orcinol)	
		15 Lectures
Unit III: Nutrition	1 <u>-</u>	
3.1 Definition: C 130mb calor BV and PEF	Calorie, Joule, Food calorimetry - calorific value determination by rimeter, calorific values of proximate principles, concept of BMI,	
3.2 BMR – defir	nition, factors affecting BMR, Significance of BMR in clinical diagnosis	
3.3 SDA/DIT -G for various	eneral concept and significance, energy requirement of individuals activities- sedentary, moderate and heavy	
3.4 Nutritional s and water	ignificance of carbohydrates, Protein, lipids, vitamins, minerals	
3.5 Formulation	of balanced diet	
3.6 Numerical p	roblems based on above concepts	

USBCH202 - Introduction to Cell biology, Physiology and Microbiology

Course Code	Title	Credits
USBCH202	Introduction to Cell biology, Physiology and Microbiology	2 Credits (45 lectures)
Unit I: Physio	ogy of digestion and absorption	15 Lectures
11 Parts and	Functions of gastro intestinal tract (GIT)	
Secretions and Bile ju	and Juices of GIT (Saliva, Gastric juice, Intestinal juice, pancreatic ce)	
1.3 Digestion	and Absorption of carbohydrates	
1.4 Digestion	and Absorption of Lipids	
1.5 Digestion	and Absorption of Proteins	
1.6 Disorders	Peptic ulcer, Lactose Intolerance	
Unit II : Physic	blogy of respiration and excretion	15 Lectures
2.1 Respirate	ry system,	
2.2 Breathing	- inspiration and expiration,	
2.3 Composi	ion of air and partial pressure of gases	
2.4 Physical 2.4.1 Transpo 2.4.2 Transpo Respirat	exchange of gases t of oxygen t of carbon dioxide ory disorders – cyanosis, respiratory acidosis and alkalosis	
 2.5 Excretion 2.5.1 Structure function, Renal tu Henle's I 2.5.2 Urine form Cl⁻ and H 	on e of the nephron: Bowman's capsule & glomerulus - Structure & (ultrafiltration, pressures involved, GFR, regulation of GFR); pule - structure & function (proximal and distal convoluted tubules and pop) nation: Reabsorption / Secretion of glucose, Na+, K+. HCO3 t ⁺ : renal threshold, Excretory disorder: Nephritis	
Unit III: Microl	biology II	15 Lectures
3.1 Microbial growth, G	Growth - Growth Curve, Mathematical expression, Synchronous eneration time	
3.2 Culture m	edia (N, C, Special requirements), Natural and Synthetic media	
3.3 Sterilizatio	n and Disinfection techniques	
3.4 Physical Autoclave Chemica	Agent of sterilization - Temperature- Pressure (Hot Air Oven,), Radiations (UV, Gamma) (examples with mechanism) I agents of sterilization - Alcohol, Halogens, Formaldehyde	

SEMESTER II – USBCHP02

PRACTICAL - I

45 Lecture

hours

45 Lecture hours

- 1. Qualitative tests for lipids
 - a) Miscibility test
 - b) Saponification test
 - c) Unsaturation test
 - d) Sudan black dye test
 - e) Salkowski test for cholesterol
- 2. Determination of SAP value of given oil sample
- 3. Determination of Acid value of give oil sample
- 4. Staining of DNA and RNA (methyl green: pyronine) using onion peel
- 5. Qualitative tests for DNA (DPA) & RNA (Orcinol) (Neumann's test for presence of phosphorus) -
- 6. Estimation of Calcium by oxalate method
- 7. Qualitative analysis for Proteins (albumin, peptone, gelatine and casein any four proteins)

SEMESTER II

PRACTICAL - II

- 1. Identification of organs / parts of digestive system
- 2. Identification of organs / parts of respiratory system
- 3. Identification of organs / parts of excretory system
- 4. Analysis of the action of salivary α amylase action on starch
- Concept of Dialysis: Ammonium sulphate precipitation → Dialysis (Test with BaCl₂ for presence of sulphate in the buffer or water outside)
- 6. Estimation of total acidity of gastric juice
- 7. Urine analysis:
 - Inorganic constituents: SO₄⁻² (BaCl₂), Cl⁻ (AgNO₃), Na⁺, K⁺ (Flame test)
 - Organic constituents: Urea, Uric acid, Creatinine
 - Abnormal constituents glucose by Benedicts method, proteins by Hellers ring test

Scheme of Examination:

Semester end assessment: It is defined as the assessment of the learners on thebasis of performance in the semester end theory/ written/ practical examination.

a) Th	eor	V	100 marks	
Que	Question Paper Pattern for Semesters I & II (100 marks)			
Q1) (Obje	ctive questions based on all units with no internal options:	20 marks	
	a) b)	Define the following (10 marks) : (Provide 5 terms to be defined, each definition will carry 2 marks) True or False with reasons (10 marks): (Provide 5 statements, for each has to state whether it is true or false and provide reasons.)	h the student	
Q2)	Que	estions based on Unit I	20 marks	
(either answer any 4 out of 8 sub-questions OR any 2 out of 4)				
Q 3)	Que	estions based on Unit II	20 marks	
(either answer any 4 out of 8 sub-questions OR any 2 out of 4)				
Q4)	Que	estions based on Unit III	20 marks	
(either answer any 4 out of 8 sub-questions OR any 2 out of 4)				
Q 5)	Que (An	stions based on Units I,II,III swer 4 out of 6 sub-questions)	20 marks	

b) Practicals 50 marks

The Course having Practical training will have Practical Examination **20 marks** for 50 marks at theend of Semester, out of which 30 marks for the Practical task assigned at the time of examination. The 20 marks are allotted as Internal Assessment.

Sr. No	Evaluation type	Marks
1	Two best practicals	10
2	Journal	05
3	Viva	05

Practical External Assessment

30 marks

Suggested Reading

- 1. De Robertis, E.D.P. and De Robertis, E.M.F. 2006. Cell and Molecular Biology. 8thedition. Lippincott Williams and Wilkins, Philadelphia.
- 2. Becker W. M. Kleinsmith, L.J., Hardin. J. and Bertoni, G. P. 2009. The World of the Cell. 7th edition, Pearson Benjamin Cummings Publishing, San Francisco
- 3. Cooper, G.M. and Hausman, R.E. 2009. The Cell: A Molecular Approach. 5th edition. ASM Press &Sunderland,Washington, D.C.; Sinauer Associates, MA.
- 4. Lehninger, Albert L, Biochemistry, Kalyani Publishers
- 5. Nelson, D. L. and Cox, M.M, (2008). Lehninger, Principles of Biochemistry 5th Edition, W. H. Freeman and Company, NY., USA.
- 6. Voet, D. and Voet, J.G. (2004) Biochemistry, 3rd Edition, John Wiley & Sons, Inc. USA.Biochemistry by Zubay, Geoffrey L.; Wm. C. Brown publishers
- 7. Zubay, Geoffrey L., Biochemistry; Wm.C.Brown publishers
- 8. Stryer, Lubert; W.H.; Biochemistry; Freeman publishers.
- 9. Harpers illustrated biochemistry by Murray, Robert K. etal.; Mc Graw Hill
- 10. Freifelder, D. (1982) Physical Biochemistry 2nd edition, W.H. Freeman and Co.NY. USA.
- 11. Cooper, T.G. (1977) TheTools of Biochemistry John Wiley and Sons, N.Y. USA.
- 12. Pattabhi. V. and Gautham N. (2002) Biophysics. Narosa Publishing House, India.
- 13. Roy, R.N. (2005) A Textbook of Biophysics. New Central Book Agency(P) Ltd., Calcutta, India
- 14. Guyton, Arthur C. and Hall, John E.;Text book of Medical physiology ; Harcourt Brace & Company Asia Pvt. Ltd.
- 15. Orten, J.M. and Neuhaus, O.W.; Human biochemistry; Mosby publishers.
- 16. Davidson, S. etal; Human nutrition and dietetics; Churchill Livingstone Publishers.
- 17. Joshi, Shubhangini A.; Nutrition and dietetics; Tata Mc Graw and Hill publishers.
- 18. Srilakshmi, B.; Nutrition Science; New Age International publishers.
- 19. Plummer, David T.; Introduction to practical biochemistry; Tata Mc. Graw and Hill publishers.
- 20. Boyer, Rodney F.Modern experimental biochemistry
- 21. Sawhney, S.K. and Singh, Randhir; Introductory practical biochemistry; Narosa Publishing House.
- 22. Verma, P.S. and Agarwal V.K.; Cell Biology, Genetics, Molecular biology, Evolution and Ecology ; Publishers : S. Chand and Co.Ltd., (2009)
- 23. Essential Cell Biology Ed: Bruce Alberts, Dennis Bray, Karen Hopkin and Alexander Johnson (2009) 3rd Edition Pub: Garland Science
- 24. B. Hall and B. HallgrimssonStrickberger's Evolution ; 4th Edition (2008) Jones and Bartlett Publishers

- 25. Sean B. Carrol, Remarkable Creatures: Epic Adventures in Search of the Origin of Species (2009). Mariner Books
- 26. Stanier, Ingraham et al ,General Microbiology 4th & 5th Ed. 1987, Macmillan Education Ltd 6.
- 27. Pelczar Michael J.; Chan Jr., E.C.S., Krieg ,Noel R.; Microbiology TMH 5th Edition .
- 28. Ananthanarayanan and Panniker, Textbook of Microbiology 5th Edition (1996). Orient Longman